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Ratio # 2: The Volume of a Mole of Gas at STP. This quantity is used specifically ONLY for particles in the gaseous state. Its value is FIXED at 22.4

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If you are asked to find out how many moles you can calculate the number of moles by remembering that grams / grams per mole = # of moles. If you've got a gas,

CHEMFILE MINI-GUIDE TO PROBLEM SOLVING CHAPTER 11 Gas Laws According to Charles s law, The ideal gas law will then allow the calculation of the number

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contains hydrogen gas. A gas cylinder contains 0.722 m³ of hydrogen gas at a presssure of 10.6 atm. A weather balloon has a maximum bolume of 7.50 x 10³ L.

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